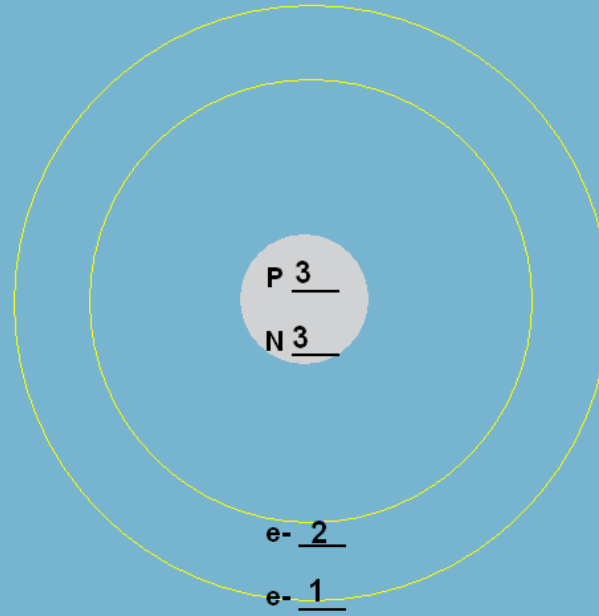


1. Fill in the blanks inside the atoms: fill in all the protons P ___ first. Your professor will help with neutrons N ____. Protons = electrons, e ___.
2. Write the chemical symbol above the atom showing atomic weight and number.
3. Write the name of the element below the atom.
4. See page 2 for an example (Lithium) filled out.

atomic weight (mass) 4
 atomic number (protons) 2 ^{He}
 Helium

| | | | | | | | | | |
|----------|-------------|-------------|-------------|-------------|-------------|-------------|-------------|-------------|-------------|
| ROW ____ | | | | | | | | | |
| ROW ____ | | | | | | | | | |
| ROW ____ | | | | | | | | | |
| | COLUMN ____ | COLUMN ____ | COLUMN ____ | COLUMN ____ | COLUMN ____ | COLUMN ____ | COLUMN ____ | COLUMN ____ | COLUMN ____ |

⁶
Li
3



Lithium